

AVGAS

Puma Energy Australia

Chemwatch: 46-4893 Version No: 3.1.1.1 Safety Data Sheet according to WHS and ADG requirements Chemwatch Hazard Alert Code: 4

Issue Date: 18/12/2014 Print Date: 08/04/2015 Initial Date: Not Available L.GHS.AUS.EN

SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

Product Identifier

Product name	AVGAS
Synonyms	Not Available
Proper shipping name	MOTOR SPIRIT or GASOLINE or PETROL (see 3.2.5 for relevant [AUST.] entries)
Other means of identification	Not Available

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses For aviation gasoline fuel piston engine aircraft. Should not be used as a solvent nor cleaning agent.

Details of the manufacturer/importer

Registered company name	Puma Energy Australia
Address	23 Theodore Street Eagle Farm 4009 QLD Australia
Telephone	1300 723 706 (24 hour contact)
Fax	1300 723 321
Website	www.Pumaenergy.com
Email	PumaAu-Safety@pumaenergy.com

Emergency telephone number

Association / Organisation	Not Available
Emergency telephone numbers	1800 039 008 (24hours)
Other emergency telephone numbers	Not Available

CHEMWATCH EMERGENCY RESPONSE

Primary Number	Alternative Number 1	Alternative Number 2
1800 039 008	+612 9186 1132	Not Available

Once connected and if the message is not in your prefered language then please dial 01

SECTION 2 HAZARDS IDENTIFICATION

Classification of the substance or mixture

HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the Model WHS Regulations and the ADG Code.

CHEMWATCH HAZARD RATINGS

	Min	Max	
Flammability	4		1
Toxicity	3		0 = Minimum
Body Contact	3		1 = Low
Reactivity	1		3 = High
Chronic	3		4 = Extreme

Poisons Schedule	S6
GHS Classification ^[1]	Flammable Liquid Category 1, Acute Toxicity (Oral) Category 3, Acute Toxicity (Dermal) Category 3, Acute Toxicity (Inhalation) Category 2, Skin Corrosion/Irritation Category 2, Germ Cell Mutagen Category 1B, Carcinogen Category 1B, Reproductive Toxicity Category 1A, STOT - SE (Narcosis) Category 3, STOT - RE Category 2, Aspiration Hazard Category 1, Acute Aquatic Hazard Category 2, Chronic Aquatic Hazard Category 2
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HSIS ; 3. Classification drawn from EC Directive 1272/2008 - Annex VI



SIGNAL WORD DANGER

Hazard statement(s)

H224	Extremely flammable liquid and vapour
H301	Toxic if swallowed
H311	Toxic in contact with skin
H330	Fatal if inhaled
H315	Causes skin irritation
H340	May cause genetic defects
H350	May cause cancer
H360	May damage fertility or the unborn child
H336	May cause drowsiness or dizziness
H373	May cause damage to organs through prolonged or repeated exposure
H304	May be fatal if swallowed and enters airways
H401	Toxic to aquatic life
H411	Toxic to aquatic life with long lasting effects

Precautionary statement(s) Prevention

P201	Obtain special instructions before use.
P210	Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.
P260	Do not breathe dust/fume/gas/mist/vapours/spray.
P270	Do not eat, drink or smoke when using this product.
P271	Use only outdoors or in a well-ventilated area.
P280	Wear protective gloves/protective clothing/eye protection/face protection.
P273	Avoid release to the environment.
P284	[In case of inadequate ventilation] wear respiratory protection.
P240	Ground/bond container and receiving equipment.
P241	Use explosion-proof electrical/ventilating/lighting/intrinsically safe equipment.
P242	Use only non-sparking tools.
P243	Take precautionary measures against static discharge.

Precautionary statement(s) Response

P301+P310	IF SWALLOWED: Immediately call a POISON CENTER/doctor/physician/first aider
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing
100411040	
P308+P313	IF exposed or concerned: Get medical advice/attention.
P330	Rinse mouth.
P331	Do NOT induce vomiting.
P370+P378	In case of fire: Use alcohol resistant foam or normal protein foam for extinction.
P302+P352	IF ON SKIN: Wash with plenty of water and soap
P361+P364	Take off immediately all contaminated clothing and wash it before reuse.
P391	Collect spillage.
P303+P361+P353	IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water/shower.
P332+P313	If skin irritation occurs: Get medical advice/attention.

Precautionary statement(s) Storage

P403+P233	Store in a well-ventilated place. Keep container tightly closed.
P403+P235	Store in a well-ventilated place. Keep cool.
P405	Store locked up.

Precautionary statement(s) Disposal

P501

Dispose of contents/container to authorised chemical landfill or if organic to high temperature incineration

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

Mixtures

CAS No	%[weight]	Name
86290-81-5	>99	gasoline
78-00-2	0-0.125	tetraethyl lead

SECTION 4 FIRST AID MEASURES

Description of first aid measures

Eye Contact	If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If skin or hair contact occurs: Quickly but gently, wipe material off skin with a dry, clean cloth. Immediately remove all contaminated clothing, including footwear. Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor.
Inhalation	 If furnes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay.
Ingestion	 IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY. For advice, contact a Poisons Information Centre or a doctor. Urgent hospital treatment is likely to be needed. In the mean time, qualified first-aid personnel should treat the patient following observation and employing supportive measures as indicated by the patient's condition. If the services of a medical officer or medical doctor are readily available, the patient should be placed in his/her care and a copy of the MSDS should be provided. Further action will be the responsibility of the medical specialist. If medical attention is not available on the worksite or surroundings send the patient to a hospital together with a copy of the MSDS. Where medical attention is not immediately available or where the patient is more than 15 minutes from a hospital or unless instructed otherwise: INDUCE vomiting with fingers down the back of the throat, ONLY IF CONSCIOUS. Lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. NOTE: Wear a protective glove when inducing vomiting by mechanical means.

Indication of any immediate medical attention and special treatment needed

Any material aspirated during vomiting may produce lung injury. Therefore emesis should not be induced mechanically or pharmacologically. Mechanical means should be used if it is considered necessary to evacuate the stomach contents; these include gastric lavage after endotracheal intubation. If spontaneous vomiting has occurred after ingestion, the patient should be monitored for difficult breathing, as adverse effects of aspiration into the lungs may be delayed up to 48 hours.

For acute or short term repeated exposures to petroleum distillates or related hydrocarbons:

- Primary threat to life, from pure petroleum distillate ingestion and/or inhalation, is respiratory failure.
- Patients should be quickly evaluated for signs of respiratory distress (e.g. cyanosis, tachypnoea, intercostal retraction, obtundation) and given oxygen. Patients with inadequate tidal volumes or poor arterial blood gases (pO2 50 mm Hg) should be intubated.
- Arrhythmias complicate some hydrocarbon ingestion and/or inhalation and electrocardiographic evidence of myocardial injury has been reported; intravenous lines and cardiac monitors should be established in obviously symptomatic patients. The lungs excrete inhaled solvents, so that hyperventilation improves clearance.
- A chest x-ray should be taken immediately after stabilisation of breathing and circulation to document aspiration and detect the presence of pneumothorax.

Wear breathing apparatus plus protective gloves in the event of a fire.

Consider evacuation (or protect in place).

Prevent, by any means available, spillage from entering drains or water course.

- Epinephrine (adrenalin) is not recommended for treatment of bronchospasm because of potential myocardial sensitisation to catecholamines. Inhaled cardioselective bronchodilators (e.g. Alupent, Salbutamol) are the preferred agents, with aminophylline a second choice.
- + Lavage is indicated in patients who require decontamination; ensure use of cuffed endotracheal tube in adult patients. [Ellenhorn and Barceloux: Medical Toxicology]

SECTION 5 FIREFIGHTING MEASURES

Fire Fighting

Extinguishing media	
	▶ Foam.
	Dry chemical powder.
	 BCF (where regulations permit).
	► Carbon dioxide.
	Water spray or fog - Large fires only.
Special hazards arising fro	om the substrate or mixture
Fire Incompatibility	Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result
Advice for firefighters	
	 Alert Fire Brigade and tell them location and nature of hazard.
	May be violently or explosively reactive.

	 Cool fire exposed containers with water spray from a protected location. If safe to do so, remove containers from path of fire.
Fire/Explosion Hazard	 Liquid and vapour are highly flammable. Severe fire hazard when exposed to heat, flame and/or oxidisers. Vapour may travel a considerable distance to source of ignition. Heating may cause expansion or decomposition leading to violent rupture of containers. On combustion, may emit toxic fumes of carbon monoxide (CO). Combustion products include; carbon dioxide (CO2) other pyrolysis products typical of burning organic material Contains low boiling substance: Closed containers may rupture due to pressure buildup under fire conditions.

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Minor Spills	 Remove all ignition sources. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb small quantities with vermiculite or other absorbent material. Wipe up. Collect residues in a flammable waste container.
Major Spills	 Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. May be violently or explosively reactive. Wear breathing apparatus plus protective gloves. Prevent, by any means available, spillage from entering drains or water course. Consider evacuation (or protect in place). No smoking, naked lights or ignition sources. Increase ventilation. Stop leak if safe to do so. Water spray or fog may be used to disperse /absorb vapour. Contain spill with sand, earth or vermiculite. Use only spark-free shovels and explosion proof equipment. Collect recoverable product with sand, earth or vermiculite. Collect solid residues and seal in labelled drums for disposal. Wash area and prevent runoff into drains. If contamination of drains or waterways occurs, advise emergency services.
	Personal Protective Equipment advice is contained in Section 8 of the MSDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe hand	ling
Safe handling	 DO NOT allow clothing wet with material to stay in contact with skin Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. Prevent concentration in hollows and sumps. DO NOT enter confined spaces until atmosphere has been checked. Avoid smoking, naked lights, heat or ignition sources. When handling, DO NOT eat, drink or smoke. Vapour may ignite on pumping or pouring due to static electricity. DO NOT use plastic buckets. Earth and secure metal containers when dispensing or pouring product. Use spark-free tools when handling. Avoid contact with incompatible materials. Keep containers securely sealed. Avoid physical damage to containers. Always wash hands with soap and water after handling. Work clothes should be laundered separately. Use good occupational work practice. Observe manufacturer's storage and handling recommendations contained within this MSDS. Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions. Contains low boiling substance: Storage in sealed containers may result in pressure buildup causing violent rupture of containers not rated appropriately. Check for bulging containers. Vent periodically Always release caps or seals slowly to ensure slow dissipation of vapours
Other information	 Store in original containers in approved flame-proof area. No smoking, naked lights, heat or ignition sources. DO NOT store in pits, depressions, basements or areas where vapours may be trapped. Keep containers securely sealed. Store away from incompatible materials in a cool, dry well ventilated area. Protect containers against physical damage and check regularly for leaks. Observe manufacturer's storage and handling recommendations contained within this MSDS.

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Conditions for safe storage, including any incompatibilities

Suitable container	 Packing as supplied by manufacturer. Plastic containers may only be used if approved for flammable liquid. Check that containers are clearly labelled and free from leaks. For low viscosity materials (i) : Drums and jerry cans must be of the non-removable head type. (ii) : Where a can is to be used as an inner package, the can must have a screwed enclosure. For materials with a viscosity of at least 2680 cSt. (23 deg. C) For maturfactured product having a viscosity of at least 250 cSt. (23 deg. C) Manufactured product that requires stirring before use and having a viscosity of at least 20 cSt (25 deg. C): (i) Removable head packaging; (ii) Cans with friction closures and (iii) low pressure tubes and cartridges may be used. Where combination packages are used, and the inner packages are of glass, there must be sufficient inert cushioning material in contact with inner and outer packages In addition, where inner packagings are glass and contain liquids of packing group I there must be sufficient inert absorbent to absorb any spillage, unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic.
Storage incompatibility	Avoid reaction with oxidising agents

PACKAGE MATERIAL INCOMPATIBILITIES

Not Available

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	gasoline	Petrol (gasoline)	900 mg/m3	Not Available	Not Available	Not Available
Australia Exposure Standards	tetraethyl lead	Tetraethyl lead (as Pb)	0.1 mg/m3	Not Available	Not Available	Sk

EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
gasoline	Gasoline	Not Available	Not Available	Not Available
tetraethyl lead	Tetraethyl lead	0.36 mg/m3	4 mg/m3	62 mg/m3
In one diama	Oniminal IDLU	D	wine of IDLU	

Ingredient	Original IDLH	Revised IDLH
gasoline	Not Available	Not Available
tetraethyl lead	40 mg/m3	40 [Unch] mg/m3

MATERIAL DATA

Exposure controls

Appropriate engineering controls	 Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use. Employees may need to use multiple types of controls to prevent employee overexposure. Mork should be undertaken in an isolated system such as a "glove-box". Employees should wash their hands and arms upon completion of the assigned task and before engaging in other activities not associated with the isolated system. Within regulated areas, the carcinogen should be stored in sealed containers, or enclosed in a closed system, including piping systems, with any sample ports or openings closed while the carcinogens are contained within. Open-vessel systems are prohibited. Each operation should be provided with continuous local exhaust ventilation so that air movement is always from ordinary work areas to the operation. Exhaust air should not be discharged to regulated areas, non-regulated areas or the external environment unless decontaminated. Clean make-up air should be introduced in sufficient volume to maintain correct operation of the local exhaust system. For maintenance and decontamination activities, authorized employees entering the area should be provided with and required to wear clean, impervious gaments, including gloves, bo
Personal protection	
Eye and face protection	 Safety glasses with side shields. Chemical goggles. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid nersonnel should be trained in their removal and suitable enument should be

readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed

Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection: AVGAS Not Available

AVGAS NOT AVAIIADI

Material

* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final

selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

CPI

Respiratory protection

Type A-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	A-AUS P2	-	A-PAPR-AUS / Class 1 P2
up to 50 x ES	-	A-AUS / Class 1 P2	-
up to 100 x ES	-	A-2 P2	A-PAPR-2 P2 ^

^ - Full-face

 $\begin{array}{l} \mathsf{A}(\mathsf{All}\ \mathsf{classes}) = \mathsf{Organic}\ \mathsf{vapours},\ \mathsf{B}\ \mathsf{AUS}\ \mathsf{or}\ \mathsf{B1} = \mathsf{Acid}\ \mathsf{gasses},\ \mathsf{B2} = \mathsf{Acid}\ \mathsf{gas}\ \mathsf{or}\ \mathsf{hydrogen}\ \mathsf{cyanide}(\mathsf{HCN}),\ \mathsf{B3} = \mathsf{Acid}\ \mathsf{gas}\ \mathsf{or}\ \mathsf{hydrogen}\ \mathsf{cyanide}(\mathsf{HCN}),\ \mathsf{E} = \mathsf{Sulfur}\ \mathsf{dioxide}(\mathsf{SO2}),\ \mathsf{G} = \mathsf{Agricultural}\ \mathsf{chemicals},\ \mathsf{K} = \mathsf{Ammonia}(\mathsf{NH3}),\ \mathsf{Hg} = \mathsf{Mercury},\ \mathsf{NO} = \mathsf{Oxides}\ \mathsf{of}\ \mathsf{nitrogen},\ \mathsf{MB} = \mathsf{Methyl}\ \mathsf{bromide},\ \mathsf{AX} = \mathsf{Low}\ \mathsf{boiling}\ \mathsf{point}\ \mathsf{organic}\ \mathsf{compounds}(\mathsf{below}\ \mathsf{65}\ \mathsf{degC}) \\ \end{array}$

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance Green or blue clear and bright highly flammable liquid with a characteristic hydrocarbon odour; does not mix with water.

Physical state	Liquid	Relative density (Water = 1)	0.710
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	>250
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	0.5-0.75 @ 40C
Initial boiling point and boiling range (°C)	40-170	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	<-40	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	HIGHLY FLAMMABLE.	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	38-49	Gas group	Not Available
Solubility in water (g/L)	Immiscible	pH as a solution	Not Available
Vapour density (Air = 1)	3-4	VOC g/L	Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Information on toxicological effects

Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may produce toxic effects. Inhalation of vapours may cause drowsiness and dizziness. This may be accompanied by narcosis, reduced alertness, loss of reflexes, lack of coordination and vertigo.

Limited evidence or practical experience suggests that the material may produce irritation of the respiratory system, in a significant number of individuals, following inhalation. In contrast to most organs, the lung is able to respond to a chemical insult by first removing or neutralising the irritant and then repairing the damage. The repair process, which initially evolved to protect mammalian lungs from foreign matter and antigens, may however, produce further lung damage resulting in the impairment of gas exchange, the primary function of the lungs. Respiratory tract irritation often results in an inflammatory response involving the recruitment and activation of many cell types, mainly derived from the vascular system.

High inhaled concentrations of mixed hydrocarbons may produce narcosis characterised by nausea, vomiting and lightheadedness. Inhalation of aerosols may produce severe pulmonary oedema, pneumonitis and pulmonary haemorrhage. Inhalation of petroleum hydrocarbons consisting substantially of low molecular weight species (typically C2-C12) may produce irritation of mucous membranes, incoordination, giddiness, nausea, vertigo, confusion, headache, appetite loss, drowsiness, tremors and anaesthetic stupor. Massive exposures may produce central nervous system depression with sudden collapse and deep coma; faalities have been recorded. Irritation of the brain and/or apnoeic anoxia may produce convulsions. Although recovery following overexposure is generally complete, cerebral micro-haemorrhage of focal post-inflammatory scarring may produce epileptiform seizures some months after the exposure. Pulmonary episodes may include chemical pneumonitis with oedema and haemorrhage. The lighter hydrocarbons may produce kidney and neurotoxic effects. Pulmonary irritancy increases with carbon chain length for paraffins and olefins. Alkenes produce pulmonary oedema at high concentrations. Liquid paraffins may produce anaesthesia and depressant actions leading to weakness, dizziness, slow and shallow respiration, unconsciousness, convulsions and death. C5-7 paraffins may also produce polyneuropathy. Aromatic hydrocarbons accumulate in lipid rich tissues (typically the brain, spinal cord and peripheral nerves) and may produce functional impairment manifested by nonspecific symptoms such as nausea, weakness, faigue and vertigo; severe exposures may produce inebriation or unconsciousness. Many of the petroleum hydrocarbons are cardiac sensitisers and may cause ventricular fibrillations.

Central nervous system (CNS) depression may include nonspecific discomfort, symptoms of giddiness, headache, dizziness, nausea, anaesthetic effects, slowed reaction time, slurred speech and may progress to unconsciousness. Serious poisonings may result in respiratory depression and may be fatal.

Inhaled

The symptoms of exposure to high vapour concentrations of benzene include confusion, dizziness, tightening of the leg muscles and pressure over the forehead followed by a period of excitement. If exposure continues, the casualty quickly becomes stupefied and lapses into a coma with narcosis. In non-fatal cases, recovery is usual. Effects of inhalation may include nausea, vomiting, headache, dizziness, drowsiness, weakness, sometimes preceded by brief periods of ataxia, staggering, weak and rapid pulse, chest pain and tightness with breathlessness, pallor, cyanosis of the lips and fingertips and tinnitus. Severe exposures may produce blurred vision, shallow rapid breathing, delirium, cardiac arrhythmias, unconsciousness, deep anaesthesia, paralysis and coma characterised by motor restlessness, tremors and hyperreflexia (occasionally preceded by convulsions). Polyneuritis and persistent nausea, anorexia, muscular weakness, headache, drowsiness, insomnia and agitation may also occur. Two to three weeks after exposure, nervous irritability, breathlessness and unusual discolouration of the skin may be evident for up to four weeks. Haemotoxicity is not usually a feature of acute exposures although anaemia, thrombocytopenia, petechial haemorrhage, and spontaneous internal bleeding have been reported. Fatal exposures may result in asphyxia, central nervous system depression, cardiac and respiratory failure and circulatory collapse; sudden ventricular fibrillation may also be fatal. Death may be lethal; pathological findings include congestion, cerebral oedema, and lung haemorrhage, renal congestion, cerebral oedema and extensive petechial haemorrhage in the brain, pleurae, pericardium, urinary tract, mucous membrane and skin. Exposure to toxic levels has also produced chromosomal damage.

	Toxic effects may result from the accidental ingestion of the material; animal exp produce serious damage to the health of the individual.	eriments indicate that ingestion of less than 40 gram may be fatal or may	
	Swallowing of the liquid may cause aspiration of vomit into the lungs with the risk of haemorrhaging, pulmonary oedema, progressing to chemical pneumonitis;		
	Signs and symptoms of chemical (aspiration) pneumonitis may include coughing, gasping, choking, burning of the mouth, difficult breathing, and bluish		
Ingestion	coloured skin (cyanosis). Ingestion of petroleum hydrocarbons may produce irritation of the pharynx, oeso	phagus, stomach and small intestine with oedema and mucosal ulceration	
	resulting; symptoms include a burning sensation in the mouth and throat. Large a	amounts may produce narcosis with nausea and vomiting, weakness or	
	fibrillation and electrocardiographic changes. Central nervous system depressio	in may also occur. Light aromatic hydrocarbons produce a warm, sharp,	
	tingling sensation on contact with taste buds and may anaesthetise the tongue. pneumonitis with pulmonary oedema and haemorrhage.	Aspiration into the lungs may produce coughing, gagging and a chemical	
	At sufficiently high doses the material may be neurotoxic (i.e. poisonous to the neurotoxic (i.e. poisonous	ervous system).	
	Skin contact with the material may produce toxic effects; systemic effects may re The material produces moderate skin irritation; evidence exists, or practical exp	sult following absorption. erience predicts, that the material either	
	 produces moderate inflammation of the skin in a substantial number of indi produces significant but moderate inflammation when applied to the health 	viduals following direct contact, and/or	
	twenty-four hours or more after the end of the exposure period.		
Skin Contact	Skin irritation may also be present after prolonged or repeated exposure; this n characterised by skin redness (erythema) and swelling (oedema) which may pr	ay result in a form of contact dermatitis (nonallergic). The dermatitis is often ogress to blistering (vesiculation), scaling and thickening of the epidermis. At	
Skill Colliact	the microscopic level there may be intercellular oedema of the spongy layer of the Repeated exposure may cause skip cracking flaking or drving following pormal	ne skin (spongiosis) and intracellular oedema of the epidermis. handling and use	
	Open cuts, abraded or irritated skin should not be exposed to this material		
	skin prior to the use of the material and ensure that any external damage is suital	tas or lesions, may produce systemic injury with narmful effects. Examine the bly protected.	
	Aromatic hydrocarbons may produce skin irritation, vasodilation with erythema a contact with the light aromatics, is unlikely due to the slow rate of permeation. Br	nd changes in endothelial cell permeability. Systemic intoxication, resulting from ranching of the side chain appears to increase percutaneous absorption.	
	The vapour when concentrated has pronounced eye irritation effects and this giv reduce exposure with available control measures, or evacuate area.	es some warning of high vapour concentrations. If eye irritation occurs seek to	
Eye	Limited evidence or practical experience suggests, that the material may cause in produce significant on the last process through the process of the produce significant on the produce signi	moderate eye irritation in a substantial number of individuals and/or may	
	exposure may cause moderate inflammation (similar to windburn) characterised	by a temporary redness of the conjunctiva (conjunctivitis); temporary	
	impairment of vision and/or other transient eye damage/ulceration may occur. The liquid may produce eye discomfort and is capable of causing temporary imp	airment of vision and/or transient eye inflammation, ulceration	
	Repeated or long-term occupational exposure is likely to produce cumulative he	alth effects involving organs or biochemical systems.	
	On the basis, primarily, of animal experiments, the material may be regarded as carcinogenic to humans. There is sufficient evidence to provide a strong presumption that human exposure to the material may result in cancer on the basis of:		
	- appropriate long-term animal studies		
	There is sufficient evidence to provide a strong presumption that human exposure to the material may result in the development of heritable genetic damage,		
	- appropriate animal studies,		
	- other relevant information		
	Harmful: danger of serious damage to health by prolonged exposure through inf Serious damage (clear functional disturbance or morphological change which m	nalation, in contact with skin and if swallowed. Nav have toxicological significance) is likely to be caused by repeated or	
	prolonged exposure. As a rule the material produces, or contains a substance which produces severe lesions. Such damage may become apparent following		
	direct application in subchronic (90 day) toxicity studies or following sub-acute (28 day) or chronic (two-year) toxicity tests. There is sufficient evidence to establish a causal relationship between human exposure to the material and subsequent developmental toxic effects in the		
	off-spring. Repeated or prolonged exposure to mixed hydrocarbons may produce narcosis	with dizziness, weakness, irritability, concentration and/or memory loss, tremor	
Chronic	in the fingers and tongue, vertigo, olfactory disorders, constriction of visual field, paraesthesias of the extremities, weight loss and anaemia and degenerative changes in the liver and kidney. Chronic exposure by petroleum workers, to the lighter hydrocarbons, has been associated with visual disturbances, damage to		
	the central nervous system, peripheral neuropathies (including numbers and paraesthesias), psychological and neurophysiological deficits, bone marrow tryicities (including hypoplasia possibly due to banzane) and banzific and renal involvement. Change demand any server to patcalaure budgesthese marrow		
	defatting which produces localised dermatoses. Surface cracking and erosion n	nay also increase susceptibility to infection by microorganisms. One	
	epidemiological study of petroleum refinery workers has reported elevations in standard mortality ratios for skin cancer along with a dose-response relationship indicating an association between routine workplace exposure to petroleum or one of its constituents and skin cancer, particularly melanoma. Other studies have		
	been unable to confirm this finding.		
	Chronic exposure to benzene may cause headache, fatigue, loss of appetite and	I lassitude with incipient blood effects including anaemia and blood changes.	
	induced aplastic anaemia include suppression of leukocytes (leukopenia), red c	ells (anaemia), platelets (thrombocytopenia) or all three cell types	
	(pancytopenia). Classic symptoms include weakness, purpura, and haemorrhag blood forming tissue. Leukaemia may develop. Occupational exposures have sho	je. The most significant toxic effect is insidious and often reversible injury to the wn a relationship between exposure to benzene and production of myelogenous	
	leukaemia. There may also be a relationship between benzene exposure and the exhibit signs of central nervous system lesions and impairment of hearing.	production of lymphoma and multiple myeloma. In chronic exposure, workers	
	Panzana baamataviaity and laukaamaganiaity involve matchalism, growth facto	r regulation evidative strass. DNA damage call regulation and apartosis	
	(Yoon et al Environmental Health Perspectives, 111, pp 1411-1420, 2003)		
	Chronic solvent innaiation exposures may result in nervous system impairment a	Ind liver and blood changes. [PAT I YS]	
	тохісіту	IRRITATION	
AVGAS	Not Available	Not Available	
	ΤΟΧΙΟΙΤΥ	IRRITATION	
	Dermal (rabbit) LD50: >1900 mg/kg ^[1]	Eye (man): 500ppm/1h moderate	
gasoline	Inhalation (guinea pig) LC50: 300 mg/L/5M ^[2]	Eye (man): 140ppm/8h mild	

Inhalation (mouse) LC50: 300 mg/L/5M^[2] Inhalation (rat) LC50: 300 mg/L/5md^[2]

	Oral (rat) LD50: >4500 mg/kg ^[1]			
	ΤΟΧΙΟΙΤΥ	IRRITATIO	DN	
tetraethyl lead	Inhalation (rat) LC50: 0.85 mg/L/1H ^[2]			
	Oral (rat) LD50: 12.3 mg/kg. ^[2]			
Legend:	1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's msds. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances			
GASOLINE	 for petroleum: This product contains benzene which is known to care neuropathic. This product contains toluene. There are indication loss. This product contains ethyl benzene and naphthale Carcinogenicity: Inhalation exposure to mice caukidney tumours which are not considered relevant thutagenicity: There is a large database of mutagive predominantly negative results. All in vivo studing vice predominantly negative results. All in vivo studing ive predominantly negative results. All in vivo studing solution exposure to mice caukidney tumours which are not considered relevant thutagenicity: There is a large database of mutagive predominantly negative results. All in vivo studing assoline vapour condensate, no adverse effects on Human Effects: Prolonged/ repeated contact main irritation and penetration by other materials. Lifetime exposure of rodents to gasoline produces in male rats as a consequence of accumulation of abnormal accumulation represents lysosomal over renal medullary tubules and necrosis. A sustained continued exposure. The alpha2-microglobulin is prot in humans. The material may cause skin irritation after prolong often characterised by skin redness (erythema) and (spongiosis) and intracellular oedema of the epide WARNING: This substance has been classified to Oral (unspec) LD50: 500 - 5000 mg/kg [Manufact. 	ause acute myeloid leukaemia and n-hex as from animal studies that prolonged ex one from which there is evidence of tumo ses liver tumours, which are not conside o humans. genicity studies on gasoline and gasoline lies in animals and recent studies in exp regnant rats to high concentrations of tolin tal neurotoxicity, on the foetus. However the foetus were observed. y cause defatting of the skin which can le carcinogenicity although the relevance to the alpha2-microglobulin protein in hyali fload and leads to chronic renal tubular regenerative proliferation occurs in epi roduced under the influence of hormona ged or repeated exposure and may proc d swelling the epidermis. Histologically t rmis.	ane which has been shown to metabolize to comp posure to high concentrations of toluene may lead urs in rodents red relevant to humans. Inhalation exposure to rat blending streams, which use a wide variety of en used humans (e.g. petrol service station attendant uene (around or exceeding 1000 ppm) can cause of in a two-generation reproductive study in rats exp ad to dermatitis and may make the skin more susc b humans has been questioned. Gasoline induces ne droplets in the male (but not female) rat kidney, cell degeneration, accumulation of cell debris, min helial cells with subsequent neoplastic transforma il controls in male rats but not in females and, mor uce a contact dermatitis (nonallergic). This form of here may be intercellular oedema of the spongy la inogenic to Humans. s a tumorigen.	ounds which I to hearing s causes dpoints and s) have shown developmental osed to exptible to kidney cancer Such eralisation of attion with e importantly, f dermatitis is iyer
TETRAETHYL LEAD	The substance is classified by IARC as Group 3: NOT classifiable as to its carcinogenicity to humans. Evidence of carcinogenicity may be inadequate or limited in animal testing. (Manifestation: Eye lachrymation; nausea or vomiting)			
Acute Toxicity	¥	Carcinogenic	ity 🗸	
Skin Irritation/Corrosion	×	Reproductiv	ity 🖌	
Serious Eye Damage/Irritation	0	STOT - Single Exposu	re 🗸	
Respiratory or Skin sensitisation	STOT - Repeated Exposure		re 🗸	
Mutagenicity	×	Aspiration Haza	rd 🖌	
		Legend:	 Data required to make classification avail Data available but does not fill the criteria Data Not Available to make classification 	able a for classification
CMR STATUS				1
SKIN	tetraethyl lead Australia Exposure Standards - Skin			Sk

SECTION 12 ECOLOGICAL INFORMATION

Toxicity

NOT AVAILABLE Ingredient Endpoint Test Duration Effect BCF Value Species Not Available Not Available Not Available Not Available Not Available gasoline Not Available tetraethyl lead

Toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment. **DO NOT** discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
tetraethyl lead	LOW (Half-life = 56 days)	LOW (Half-life = 0.93 days)

Bioaccumulative potential

Ingredient	Bioaccumulation
tetraethyl lead	HIGH (BCF = 18140)
Mobility in soil	
Ingredient	Mobility
tetraethyl lead	LOW (KOC = 757.6)

SECTION 13 DISPOSAL CONSIDERATIONS

Waste treatment methods

Product / Packaging disposal	 Containers may still present a chemical hazard/ danger when empty. Return to supplier for reuse/ recycling if possible. Otherwise: If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill. Where possible retain label warnings and MSDS and observe all notices pertaining to the product. Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked. A Hierarchy of Controls seems to be common - the user should investigate: Reduction Reuse Recycling Disposal (if all else fails) This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate. D NOT allow wash water from cleaning or process equipment to enter drains. It may be necessary to collect all wash water for treatment before disposal. In all cases disposal to sever may be subject to local laws and regulations and these should be considered first. Where in doubt contact the responsible authority. Recycle wherever possible. Consult manufacture for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified. Dispose of by: burial in a land-fill specifically licenced to accept chemical and / or pharmaceutical wastes or Incinerat
	 Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.

SECTION 14 TRANSPORT INFORMATION

Labels Required



Land transport (ADG)

UN number	1203		
Packing group	II.		
UN proper shipping name	MOTOR SPIRIT or GASOLINE or PETROL (see 3.2.5 for relevant [AUST.] entries)		
Environmental hazard	No relevant data		
Transport hazard class(es)	Class 3 Subrisk Not Applicable		
Special precautions for user	Special provisions 243 363 Limited quantity 1 L		

Air transport (ICAO-IATA / DGR)

UN number	1203
Packing group	II
UN proper shipping name	Gasoline; Motor spirit; Petrol
Environmental hazard	No relevant data

	ICAO/IATA Class	3	
Transport hazard class(es)	ICAO / IATA Subrisk	Not Applicable	
	ERG Code	ЗН	
	Special provisions		A100
Special precautions for user	Cargo Only Packing Instructions		364
	Cargo Only Maximum Qty / Pack		60 L
	Passenger and Cargo Packing Instructions		353
	Passenger and Cargo Maximum Qty / Pack		5 L
	Passenger and Cargo Limited Quantity Packing Instructions		Y341
	Passenger and Cargo	Limited Maximum Qty / Pack	1 L

Sea transport (IMDG-Code / GGVSee)

UN number	1203		
Packing group	II.		
UN proper shipping name	MOTOR SPIRIT or GASOLINE or PETROL		
Environmental hazard	Not Applicable		
Transport hazard class(es)	IMDG Class 3 IMDG Subrisk Not Applicable		
Special precautions for user	EMS NumberF-E, S-ESpecial provisions243 363Limited Quantities1 L		

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

gasoline(86290-81-5) is found on the following regulatory lists	"Australia Exposure Standards","International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs","Australia Inventory of Chemical Substances (AICS)", "Australia Hazardous Substances Information System - Consolidated Lists"
tetraethyl lead(78-00-2) is found on the following regulatory lists	"Australia Exposure Standards", "International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs", "Australia Inventory of Chemical Substances (AICS)", "Australia Hazardous Substances Information System - Consolidated Lists"

National Inventory	Status
Australia - AICS	Y
Canada - DSL	Υ
China - IECSC	Y
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (gasoline; tetraethyl lead)
Korea - KECI	Υ
New Zealand - NZIoC	Y
Philippines - PICCS	Υ
USA - TSCA	Y
Legend:	Y = All ingredients are on the inventory $N = Not$ determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 OTHER INFORMATION

Other information

Ingredients with multiple cas numbers

Name	CAS No
gasoline	8006-61-9, 86290-81-5

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chernwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net/references

The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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Version No: 3.1.1.1	AVGAS	Print Date: 08/04/2015

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